

How does tropospheric OH variability influence the transport of SO₂ from the surface to the stratosphere? – A conceptual study

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Introduction

- 4-7%/year increase in strat. sulfate aerosols (1990-2009)¹
- No volcanoes → transport of carbonyl sulfide (COS) and sulfur dioxide (SO₂) via tropical tropopause layer (TTL) maintains aerosol layer
- COS → low reactivity and solubility in the troposphere
- SO₂ → lifetime largely determined by the hydroxyl (OH) concentration
- SO₃ → highly soluble, reasonably efficiently removed
- Non-soluble fraction of S compounds determines flux to stratosphere
- Last removal of soluble species at Lagrangian cold point (LCP)

What we did

- ATLAS² model → impact of trop. OH on strat. SO₂ flux
- Back trajectories → start at 400K, 2°x 2° long/lat grid, 30°N to 30°S, 120 days starting on 31 January 2010, must go to 800 hPa (3460)
- Troposphere → vertical winds used for vertical motion
- Upper TTL and strat → full sky heating rates for vertical motion
- Winds and heating rates from ECMWF reanalysis data – ERA Interim
- Latitude/altitude OH fields from GEOS-Chem tropospheric Chemistry Transport Model³ (Figure 2), provide OH along the trajectories
- Box model → run from 800 hPa to LCP to represent OH oxidation:

$$\text{SO}_2 + \text{OH} + \text{M} \rightarrow \text{HSO}_3 + \text{M} \quad (\text{R1})$$
 - R1 is the dominant gas-phase reaction
 - HSO₃ → reacts with oxygen to produce SO₃
 - SO₃ → with water vapour, converted rapidly to sulfuric acid (H₂SO₄)
 - H₂SO₄ → uptake into water droplets major source of trop sulfate
 - SO₂/SO₃/HSO₃/H₂SO₄ gas to particle conversion, i.e. microphysical processes not considered in box model → conceptual study at this point

Findings and outlook

- High OH scenarios** → 2-4% of air masses contribute >10% of initial SO₂ to strat.
- OH hole** → ~20% of air masses deliver >10% of initial SO₂. On average, 3 times more SO₂ reaches stratosphere than under the **high OH scenarios**
- Southwest and Central Pacific, Southeast Asia, and the Indian Ocean are the main contributing regions to stratospheric SO₂
- Low OH** increases mean SO₂ lifetime by 5 days in trop. and TTL
- Temperature dependence of (R1) reaction rate leads to an increase in SO₂ lifetime by ~2 days from low to high temp.
- Outlook → include microphysics as ~50% of strat. sulfur delivered as aerosol

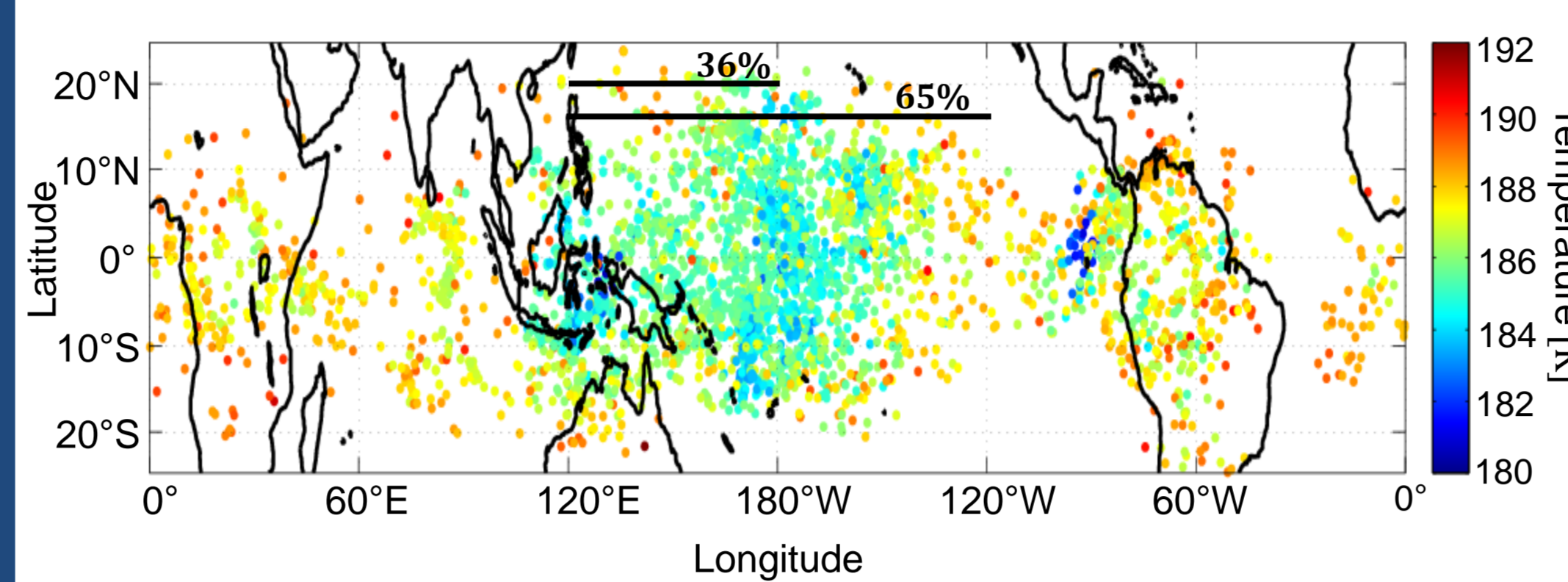


Figure 1: LCP locations and corresponding minimum temperatures. West Pacific is main strat. entrance region for trop. air masses (36%). 2009-2010 El Niño event → 29% of trajectories reach LCP above Central Pacific.

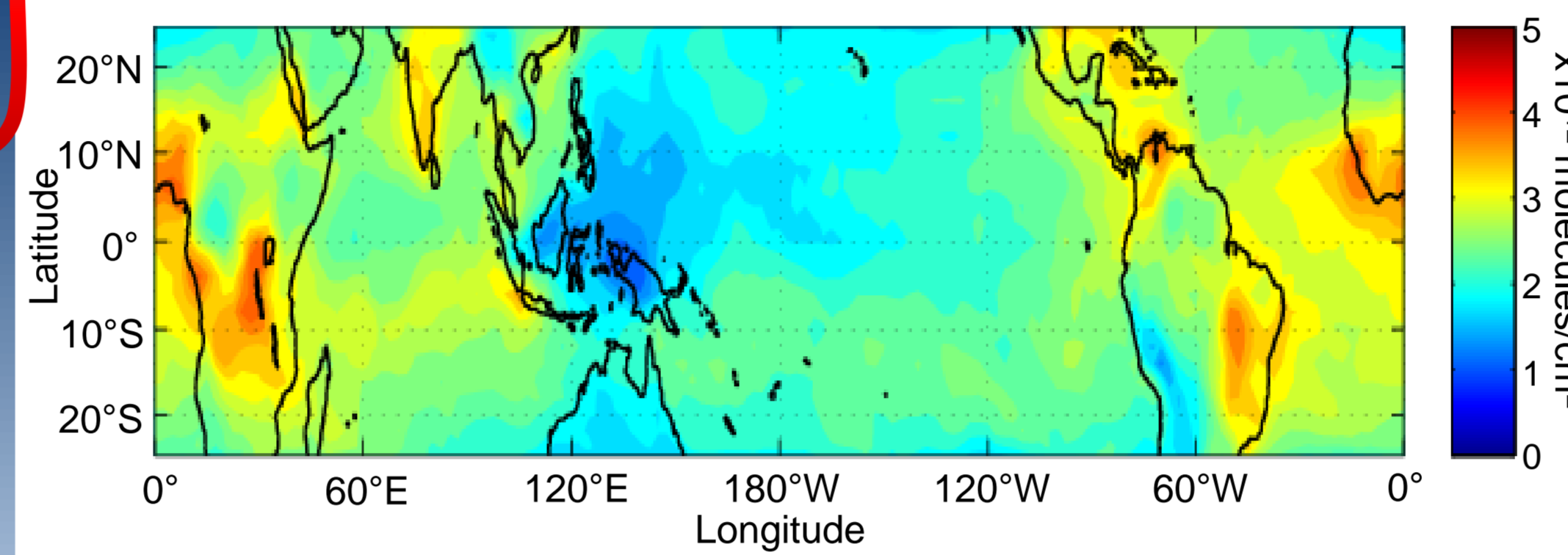


Figure 2: Trop. OH columns from GEOS-Chem (1-15 October 2009). Four scenarios considered:

- OH zonal mean → OH 'global' scenario
- OH mean from 125°E-140°E → OH 'hole' scenario
- OH mean from 50°W-50°E → OH 'Atlantic' scenario
- OH mean from 80°W-100°E → OH 'without Pacific'

- Studies of the flux of species often based on **zonal mean OH concentrations** (i)⁴
- recent study by Rex et al.⁵ showed existence of pronounced **minimum in OH-concentrations** above the West Pacific → impact of such variability in trop. OH concentrations on strat. SO₂ flux analysed here

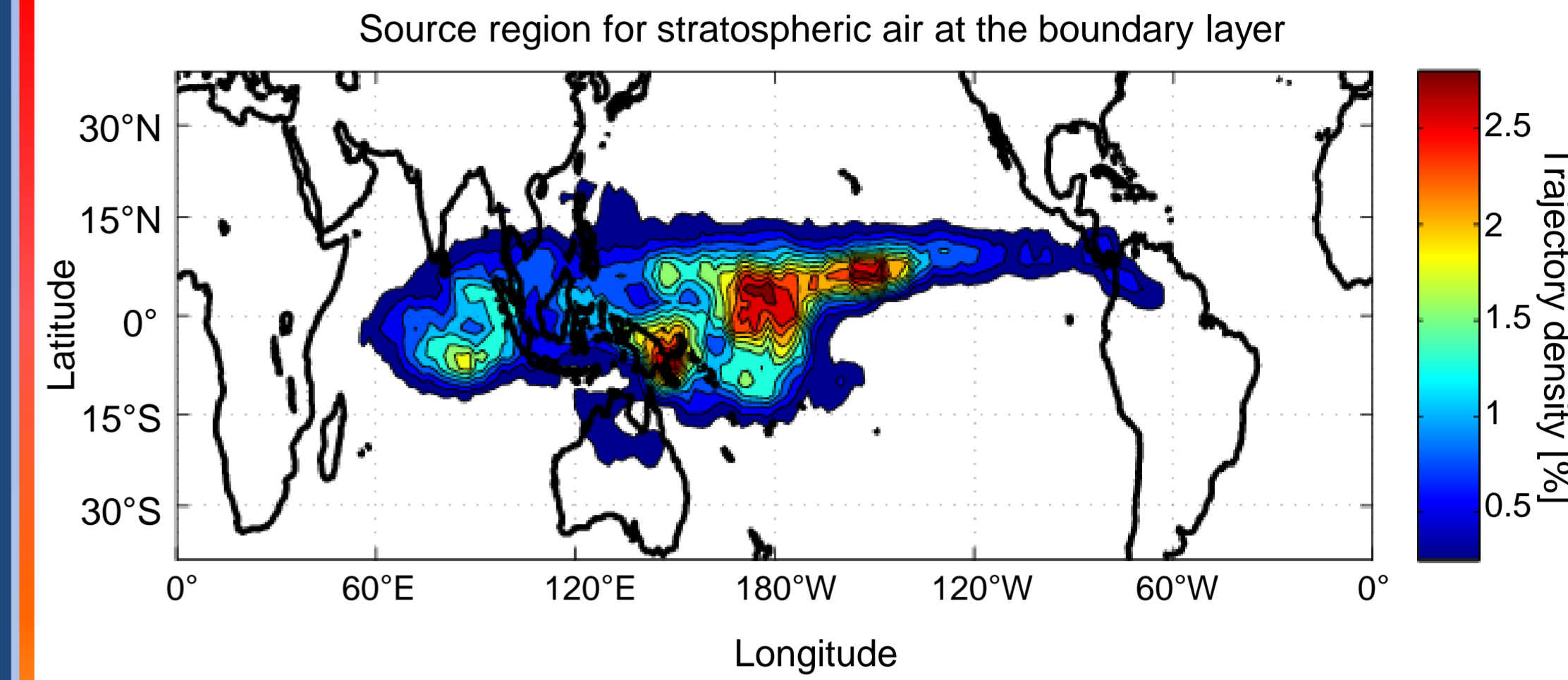


Figure 4: 800 hPa trajectory origins. Biomass burning and small volcanic eruptions in SW Pacific, SE Asia, and Central Pacific can contribute more to strat. sulfur budget than previously thought.

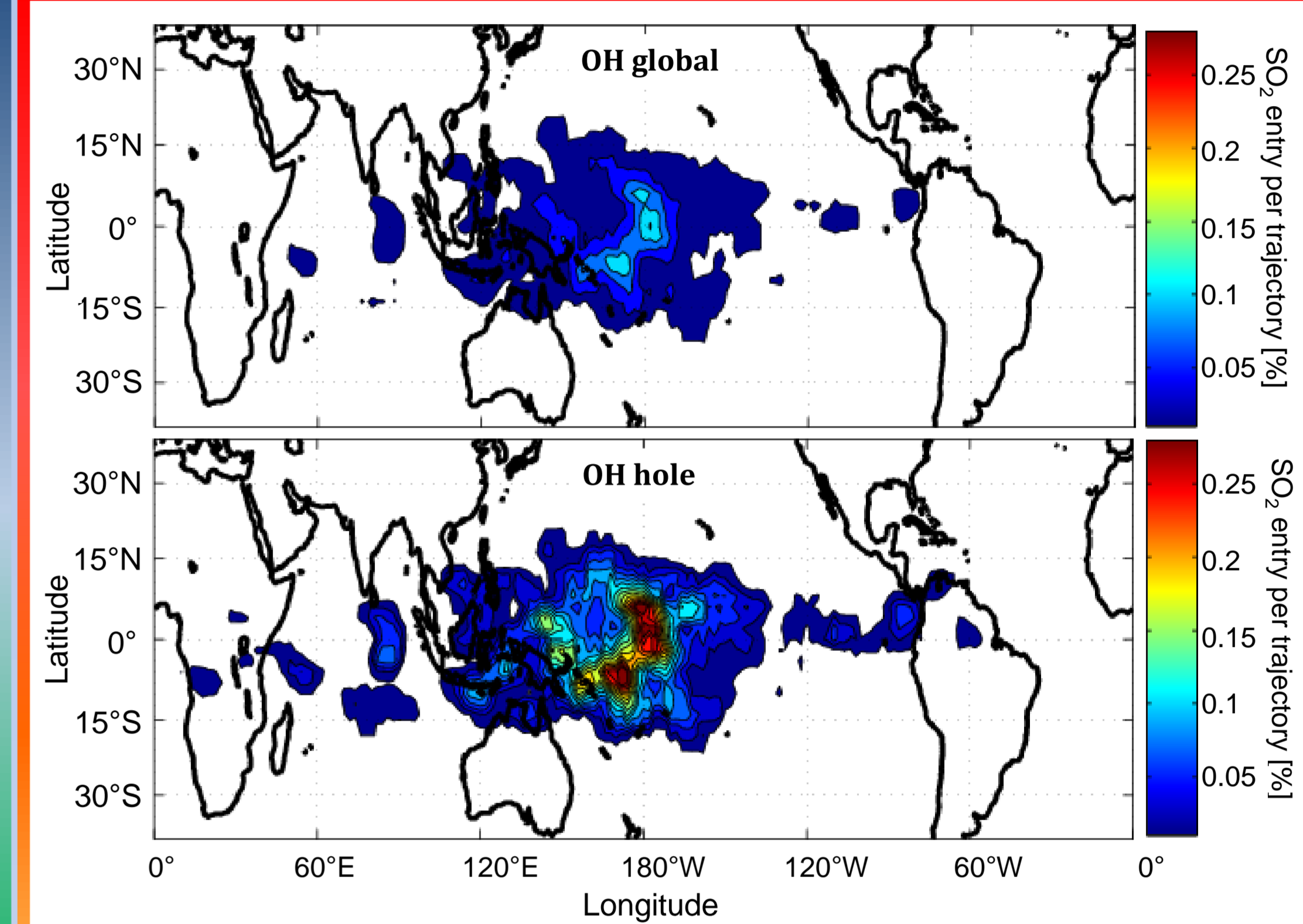


Figure 5: Regional contribution to the total % of SO₂ reaching the LCP. 10°long x 5°lat regions.

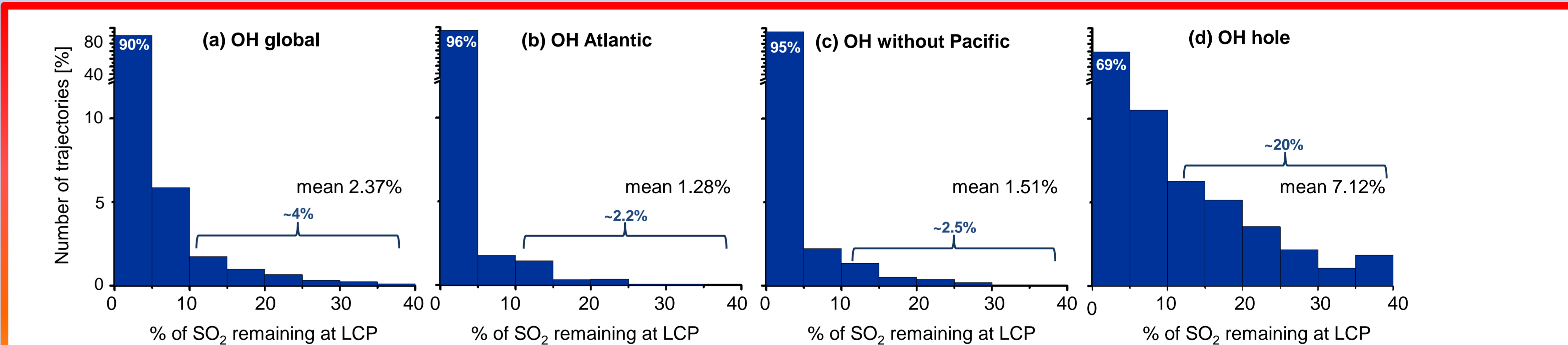


Figure 3: SO₂ remaining at LCP for 4 OH scenarios. **High OH** → ~2-4% of all trajectories deliver more than 10% of their initial SO₂ to the stratosphere. **Low OH** → ~20% of all trajectories deliver more than 10% of SO₂ to the stratosphere.

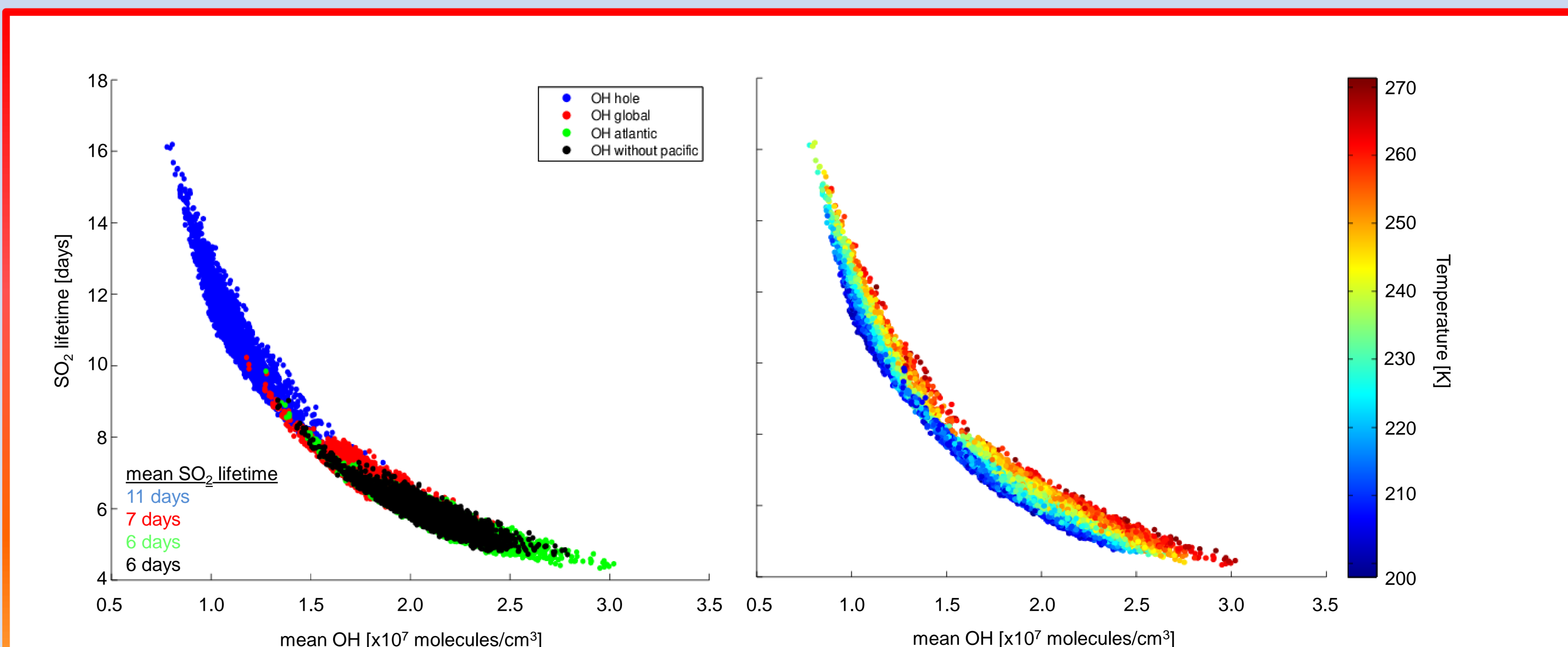


Figure 6: SO₂ lifetime vs. mean OH concentration along all trajectories for 4 OH scenarios. **Low OH** → lifetime can be twice that for **high OH**. Panel b: OH weighted mean temperature along each trajectory. Spread in the SO₂ lifetime related to temperature dependence of OH oxidation (R1).

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References

- Hofmann, D. et al.: Increase in background stratospheric aerosol observed with lidar at Mauna Loa Observatory and Boulder, Colorado, *Geophys. Res. Lett.*, 36, 2009.
- Wohltmann, I. and M. Rex: The Lagrangian chemistry and transport model ATLAS: validation of advective transport and mixing, *Geosci. Model Dev.*, 2, 153-173, 2009.
- Ridder, T., et al.: Ship-borne FTIR measurements of CO and O₃ in the Western Pacific from 43N to 35S: an evaluation of sources, *Atmos. Chem. Phys.*, 12, 815-828, 2012.
- Weisenstein et al.: A two dimensional model of sulfur species and aerosols, *J. Geophys. Res.*, 102(D11), 13019-13035, 1997.
- Rex, M. et al.: Is There a Hole in the Global OH Shield Over the Tropical Western Pacific Warm Pool?, *IUGG*, Melbourne, 2011.